- 1. The solubility of ibuprofen in an acetate buffer solution (pH 4.5) obtained by shaking in a full flask for 24 hours at 37°C) is 0.084 mg/ml, and the maximum dose registered in the Russian Federation is 400 mg. How can the solubility of ibuprofen in an acetate buffer be characterized?
- 2. There are data in the literature on the solubility of ketoprofen in water at room temperature (0.010 mg/ml), as well as at 37 °C in solutions with a pH value of 1.2 (0.13 mg/ml), pH 4.6 (0 .49 mg/ml) and pH 6.8 (40.76 mg/ml). The maximum dosage of ketoprofen in immediate release dosage forms for internal use, registered for medical use in the Russian Federation, is 100 mg. Determine the solubility of ketoprofen in the physiological pH range.
- 3. Distribute the API according to the BCS and BDDCS classification:

  Amlodipine has high solubility in physiological environments of the gastrointestinal tract, high permeability and intensive metabolism, lincomycin low permeability, high solubility, weak metabolism, warfarin dose number 2.2, high permeability, intensive metabolism, vancomycin dose number 0.02, low permeability, weak metabolism, theophylline dose number 0.3, high permeability, intensive metabolism, roxithromycin dose number 12, low permeability, weak metabolism, pilocarpine high permeability, high solubility, intensive metabolism, nystatin dose number 0.2, low permeability, poor metabolism, digoxin dose number 0.001, low permeability, weak metabolism, ampicillin dose number 0.3, low permeability, low metabolism, capsaicin dose number 12, high permeability, intensive metabolism, melatonin dose number 0.2, high permeability, intensive metabolism, melatonin dose number 0.2, high permeability, intensive metabolism.